

Chapter 8 Chemistry Test

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A)8, 2, and 3 B)7, 1, and 2 C)2, 7, and 4 D)2, 4, and 6 E)1, 5, and 7 6) Lattice energy is _____. A)the sum of electron affinities of the components in an ionic solid B)the energy required to convert a mole of ionic solid into its constituent ions in the gas phase C)the sum of ionization energies of the components in an ionic solid

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$\text{KOH(aq)} + \text{HCl(aq)} \rightarrow \text{KCl(aq)} + \text{H}_2\text{O(l)}$ - A salt is a compound formed when the hydrogen ion, H^+ from an acid is replaced by a metal ion or an ammonium ion, NH_4^+ . Salts Soluble Insoluble

Chapter 8: Salts - Pasti Skor

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