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Modern Chemistry Test B Answer

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Modern Chemistry Chapter Test B Answer Key

Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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Modern Chemistry Chapter 6 Test B Answer Key

Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

Modern Chemistry Chapter 3 Test B Answers

Modern Chemistry 110 Chapter Test Name Class Date Chapter Test B, continued ____ 7.

Effervescence is the a. dissolving of a gas in a liquid. b. escape of a liquid from a liquid-liquid solution. ... Modern Chemistry 229 Answer Key TEACHER RESOURCE PAGE. Title: mc06sete_FMct_i-iv.qxd

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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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Device b measures mass, which does not change with location because gravity affects both the measured body and the mass standard equally. Device c measures volume of a liquid, which

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changes slightly with temperature and pressure. Weight, mass, and volume do not change with the shape of the object. MODERN CHEMISTRY MEASUREMENTS AND CALCULATIONS 11

2 Measurements and Calculations

PROBLEMS Write the answers to the questions on the line to the left, and show your work in the space provided. 25. 26. MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this iso- tope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11 CHAPTER 3 TEST

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Modern Chemistry 102 Chapter Test Name Class Date Chapter Test B, continued PART IV Write the answers to the following questions on the line to the left, and show your work in the space provided. ____30. Convert 0.75 atm to mm Hg. ____31. The pressure of a sample of helium in a 200. mL container is 2.0 atm.

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APPENDIX D Problem Bank

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B. Problems Write your answer in the space provided. Date Class 11. List the five traditional branches of chemistry and explain what each involves. 12. Discuss the role of chemistry in one of these areas: Materials, Energy, Medicine and Biotechnology, Agriculture, the Environment, or the Universe.

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