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Chemistry Periodic Assessment 2 Answer Key

Unit 2: Chemistry 1. 2.2 ATOMIC STRUCTURE AND THE PERIODIC TABLE (a) atoms containing a positively charged nucleus with orbiting negatively charged electrons (b) atomic nuclei containing protons and neutrons (c) the relative masses and relative charges of protons, neutrons and electrons (e) the terms atomic number, mass number and isotope

Revising the periodic table using a concept map | 14-16

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Chemistry Periodic Assessment 2 Answer Key

22) Metals outnumber nonmetals on the periodic table because most of the elements a) are close to having a full octet b) have 1 to 2 valence electrons c) have a high ionization energy d) have a high electronegativity 23) As we move from left to right on the periodic table, the reactivity of metals a) increases b) decreases c) stays the same

a) 1, 2, and 14 b) 1, 2, and 18 c) 1, 2, and 17 d) 2, 13 ...

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SHORT ANSWER. 117. Their locations in the periodic table indicate that phosphorus and sulfur are nonmetals and sodium is a metal. Nonmetals are a group with characteristic properties, so phosphorus and sulfur are the most similar elements of the three. 118. In the periodic table, elements in the same column or group have similar properties.

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